

## Fluorescence decay time distribution analysis of cyclic enkephalin analogues; Influence of solvent and Leu configuration in position 5 on conformation<sup>★</sup>✉

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**Lifetime distribution analysis were performed to study the influence of Leu configuration in position 5 on changes of the peptide chain of cyclic analogues of enkephalins containing a fluorescence donor and acceptor in different solvents. The configuration change of Leu5 in all the analogues of enkephalins studied which contain donor-acceptor pairs has no apparent influence on Trp lifetime distributions. In contrast, there is a significant solvent effect on the shape of lifetime distribution.**

The opioid system plays a crucial role in analgesic action. The main elements of this system are endogenous peptides (enkephalins, endomorphins,  $\beta$ -endorphin, dynorphins) which are the natural ligands for opioid receptors ( $\mu$ ,  $\delta$ ,  $\kappa$ ). The  $\mu$ -opioid receptors have a great significance in an-

algesic action. Structure-biological activity analysis of opioid peptide analogues has shown that rigidifying the peptide chain (which is accomplished by cyclization) results in an increase of their affinity for opioid receptors, probably by stabilizing the bioactive conformation(s) [1, 2]. Mu-

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**Abbreviations:** MeOH, methanol; MeCN, acetonitrile; Me<sub>2</sub>SO, dimethyl sulfoxide; Dns, 5-dimethylamino-naphthalene-1-sulfonyl (dansyl); Phe(p-NO<sub>2</sub>), *p*-nitrophenylalanine; Dab,  $\alpha,\gamma$ -diaminobutyric acid; Dns-[L-EN], Dns-*cyclo*[D-Dab<sup>2</sup>-Gly<sup>3</sup>-Trp<sup>4</sup>-Leu<sup>5</sup>]; Dns-[D-EN], Dns-*cyclo*[D-Dab<sup>2</sup>-Gly<sup>3</sup>-Trp<sup>4</sup>-D-Leu<sup>5</sup>]; F(NO<sub>2</sub>)-[L-EN], Phe(p-NO<sub>2</sub>)<sup>1</sup>-*cyclo*[D-Dab<sup>2</sup>-Gly<sup>3</sup>-Trp<sup>4</sup>-Leu<sup>5</sup>]; F(NO<sub>2</sub>)-[D-EN], Phe(p-NO<sub>2</sub>)<sup>1</sup>-*cyclo*[D-Dab<sup>2</sup>-Gly<sup>3</sup>-Trp<sup>4</sup>-D-Leu<sup>5</sup>]; Fmoc, fluorene-9-yl-methoxycarbonyl; HBTU, 2-(1H-benzotriazol-1-yl)-1,1,3,3-tetramethyluronium hexafluorophosphate; <sup>1</sup>H NMR COSY, hydrogen nuclear magnetic resonance correlation spectroscopy; RP-HPLC, reversed phase high performance liquid chromatography; FAB-MS, fast atom bombardment mass spectrometry.



used [10, 16–18]. An alternative approach is not to assume any functional form of  $\alpha(\tau)$ , but to determine it from the experimental data. This approach is superior in that it makes no assumptions about the shape of the distribution. Two general algorithms: the maximum entropy method [11, 12, 19–22] and the exponential series methods are used for such calculations [23, 24]. For time-correlated single photon counting the measured fluorescence decay  $F(t)$  is a convolution of the instrument response function  $R(t)$  and the intensity decay function  $I(t)$ :

$$F(t) = \int_0^t R(s+\delta)I(t-s)ds \quad (3)$$

where  $\delta$  is the time shift parameter which takes care of the experimental or computational artifacts which cause an artificial time-shift of the calculated function with respect to experimental data.  $I(t)$  is the theoretical intensity decay function, a continuous distribution of lifetimes, as described in equation (2).  $\alpha(\tau)$  is the distribution function which is to be determined. The quality of the fit to experimental data is checked by the  $\chi^2$  statistics:

$$\chi^2 = (1/M) \sum_{i=1}^N \{F_c(t_i) - F_e(t_i)\}^2 / \sigma_i^2 \approx 1.0 \quad (4)$$

where  $F_c(t_i)$  is the calculated fluorescence decay at time  $t_i$ ,  $F_e(t_i)$  is the measured decay at time  $t_i$ ,  $\sigma_i$  is the standard deviation for the  $i$ -th data point and  $M$  is the number of the degrees of freedom. Usually the good criterion of  $\chi^2 \approx 1.0$  could be obtained for many different distributions of  $\alpha(\tau)$ . The optimum is the one which fits data adequately ( $\chi^2 \approx 1.0$ ) and maximizes the value of the Shannon-Jaynes entropy function  $S$ , as defined below:

$$S = - \sum p_i \log p_i \quad (5)$$

where  $p_i = \alpha_i / \sum \alpha_i$ . If there is a priori knowledge about the distribution ( $m_i$ ), equation (5) is modified as follows:

$$S = - \sum p_i \log(p_i / m_i) \quad (6)$$

If the  $\chi^2$  criterion is satisfied for many distributions, then the maximum entropy criterion selects

that distribution which contains the minimum number of peaks of maximal width in the distribution.

## MATERIALS AND METHODS

**Synthesis.** Linear precursors of peptides (Dns-[L-EN], Dns-[D-EN], F(NO<sub>2</sub>)-[L-EN], F(NO<sub>2</sub>)-[D-EN]) were synthesized by solid-phase methodology using the Fmoc chemistry [25]. The cyclization was performed using HBTU [26]. The peptides were purified by means of the preparative RP-HPLC. The homogeneity and molecular constitution of the compounds were assessed by analytical RP-HPLC and FAB-MS, and <sup>1</sup>H-NMR COSY spectroscopy.

**Time-resolved fluorescence measurements.** Fluorescence decay times were measured using a time correlated single-photon counting apparatus at the Laboratory of Ultrafast Laser Spectroscopy, Adam Mickiewicz University (Poznań, Poland). The excitation source ( $\lambda_{\text{ex}} = 280$  nm) was a pico/femtosecond laser system (Ti:Sapphire "Tsunami" laser pumped with an argon ion laser "BeamLok" 2060) [27]. The emission was detected with a magic angle polarizer at an emission wavelength of  $\lambda_{\text{em}} = 340$  nm. A Ludox solution was used to collect the instrument response. All measurements were performed at 20°C in water, methanol, acetonitrile and Me<sub>2</sub>SO. For lifetime distribution calculation the software provided by Edinburgh Analytical Instruments was used.

## RESULTS AND DISCUSSION

If the donor and the acceptor remain at a fixed distance the decay of the donor is accelerated by energy transfer, without changing its mono-exponential time dependence. However, if there is a range of donor-acceptor distances, then a range of transfer rates, and hence a distribution of decay times should be observed. If two unique distance distributions are present one should expect a bimodal time distribution. To check how the solvent and different Leu configuration in position 5 influence the distance between the donor in posi-

tion 4 (Trp) and the acceptor (Dns or Phe(NO<sub>2</sub>)) in position 1 in the cyclic analogues of enkephalin considered in this study we measured the fluorescence decay times of the donor in the presence of the acceptor in four different solvents (H<sub>2</sub>O, MeOH, MeCN, Me<sub>2</sub>SO). Fluorescence lifetime

tor pairs has no apparent influence on the Trp lifetime distributions. In contrast, there is a significant solvent effect on the shape of the lifetime distribution. In less polar solvents (MeOH, MeCN) the distributions are asymmetric, uni- or bimodal; in bimodal distributions the long-life-

**Table 1. Lifetimes distribution parameters of Trp fluorescence (average fluorescence lifetime decays ( $\tau$ ) and amplitudes ( $f$ ) of particular components) of cyclic enkephalin analogues in different solvents**

Analogue	Lifetime [ns]	Amplitude	Lifetime [ns]	Amplitude	Lifetime [ns]	Amplitude	$\chi^2_R$
	$\tau_1$		$f_1$		$\tau_2$		
H <sub>2</sub> O							
Dns-[L-EN]	0.07	0.5903	0.25	0.3847	1.17	0.0250	1.08
Dns-[D-EN]	0.10	0.6045	0.24	0.3737	1.03	0.0217	1.04
F(NO <sub>2</sub> )-[L-EN]	0.16	0.2118	1.15	0.7782	4.66	0.0100	1.17
F(NO <sub>2</sub> )-[D-EN]	0.50	0.9136	1.31	0.0864			1.06
MeOH							
Dns-[L-EN]	0.17	0.9811	1.94	0.0189			1.02
Dns-[D-EN]	0.15	0.9941	1.73	0.0059			1.12
F(NO <sub>2</sub> )-[L-EN]	0.71	1.00					1.09
F(NO <sub>2</sub> )-[D-EN]	0.67	1.00					1.18
MeCN							
Dns-[L-EN]	0.14	0.9893	1.36	0.0078	5.83	0.0029	1.07
Dns-[D-EN]	0.10	0.9972	1.67	0.0028			1.07
F(NO <sub>2</sub> )-[L-EN]	0.61	1.00					1.12
F(NO <sub>2</sub> )-[D-EN]	0.55	1.00					1.11
Me <sub>2</sub> SO							
Dns-[L-EN]	0.22	0.9559	1.00	0.0315	5.58	0.0125	1.15
Dns-[D-EN]	0.22	0.8680	0.66	0.1241	5.88	0.0079	1.07
F(NO <sub>2</sub> )-[L-EN]	0.62	0.5561	1.19	0.4327	4.88	0.0101	1.29
F(NO <sub>2</sub> )-[D-EN]	0.56	0.3073	1.31	0.6667	3.08	0.0260	1.17

distribution parameters for cyclic enkephalin analogues containing the energy donor (Trp) and the acceptors (Dns or Phe(NO<sub>2</sub>)) calculated on the basis of the measured fluorescence intensity decay are presented in Table 1 and in a graphical form in Figs. 2 and 3. The presence of the energy acceptor results in a significant decrease of the fluorescence lifetime of tryptophan compared to the isolated donor [28].

The configuration of Leu<sup>5</sup> in all the analogues of enkephalins studied which contain donor-acceptor

time components have small amplitudes (0.3–2%). The width of the distributions is higher in the dansyl containing enkephalin analogues. In more polar solvents the number of distribution components increases. In Dns-[L or D-EN] they are narrow and well separated. The short-lifetime components have a higher amplitude than the long-lifetime ones. The conformational space of the peptides studied in MeCN and MeOH can be described in terms of a single conformational family, in which the side chains of the aromatic amino

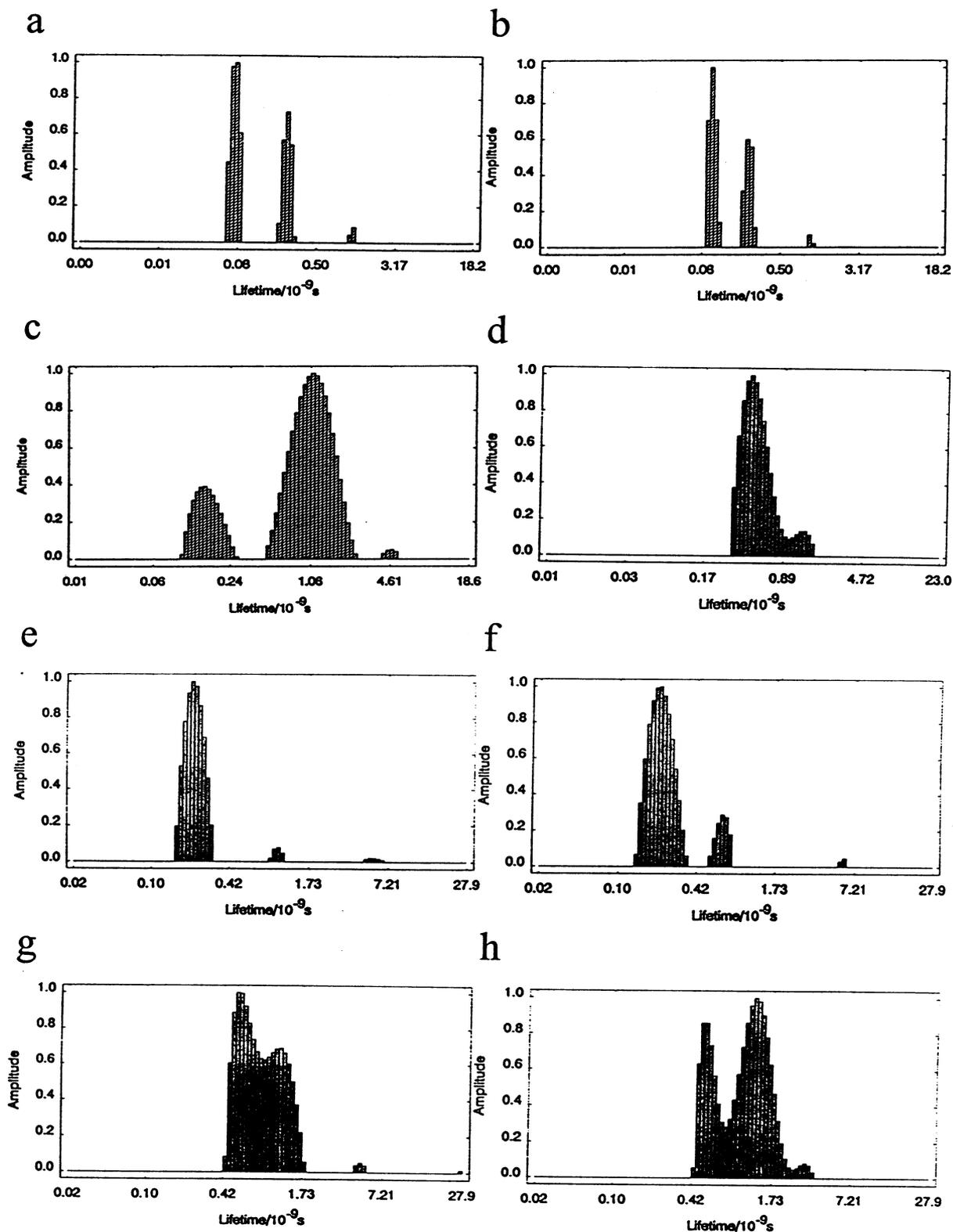


Figure 2. Lifetime distributions of the Trp fluorescence of Dns-[L-EN] (a), Dns-[D-EN] (b), F(NO<sub>2</sub>)-[L-EN] (c), F(NO<sub>2</sub>)-[D-EN] (d) in water and Dns-[L-EN] (e), Dns-[D-EN] (f), F(NO<sub>2</sub>)-[L-EN] (g), F(NO<sub>2</sub>)-[D-EN] (h) in Me<sub>2</sub>SO.

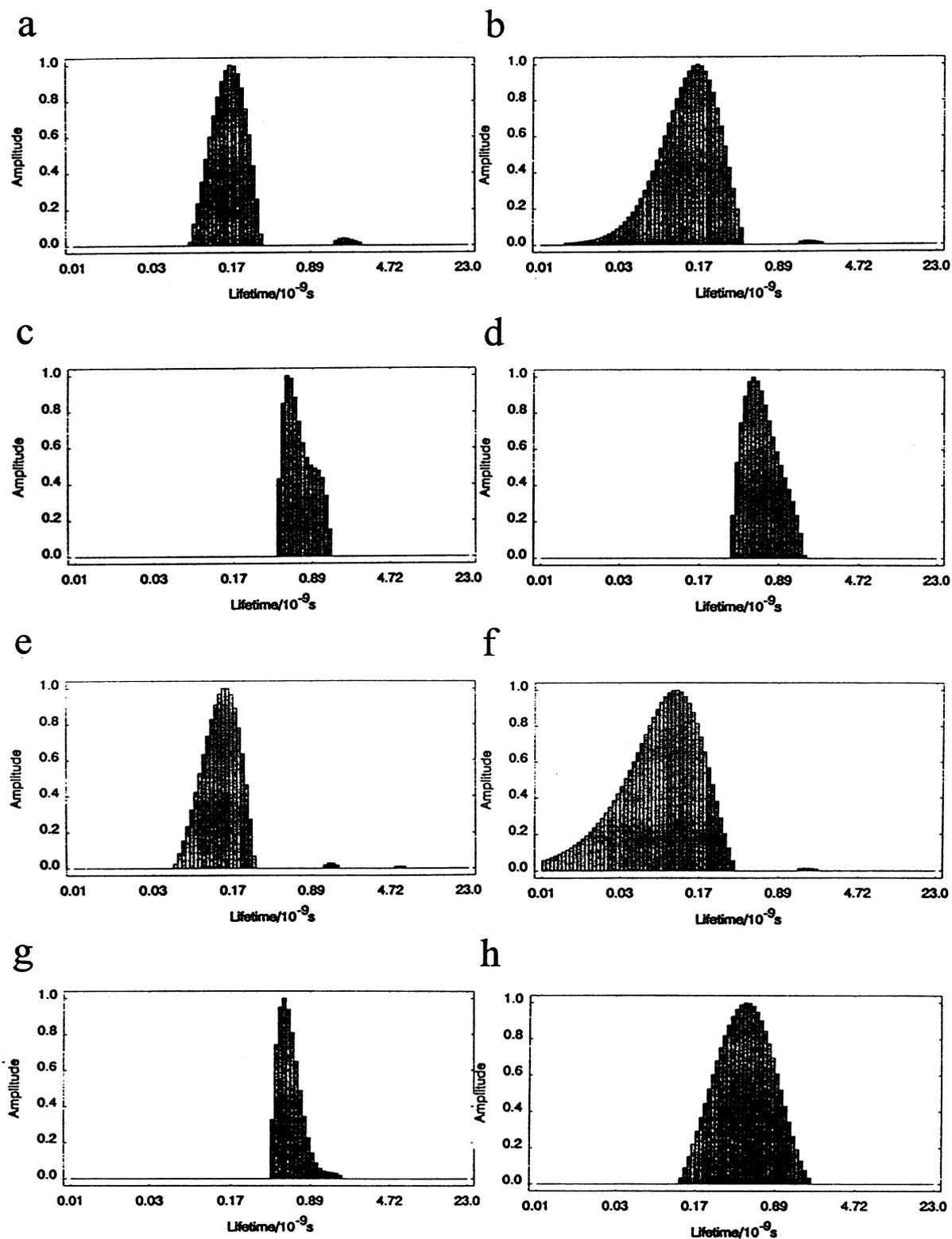


Figure 3. Lifetime distributions of the Trp fluorescence of Dns-[L-EN] (a), Dns-[D-EN] (b), F(NO<sub>2</sub>)-[L-EN] (c), F(NO<sub>2</sub>)-[D-EN] (d) in MeOH and Dns-[L-EN] (e), Dns-[D-EN] (f), F(NO<sub>2</sub>)-[L-EN] (g), F(NO<sub>2</sub>)-[D-EN] (h) in MeCN.

acids in positions 1 and 4 are highly mobile, especially in the dansyl analogues. In solvents more polar than MeOH or MeCN the number of conformational families increases. The ability of water molecules to break the intramolecular hydrogen bonds increases the conformational freedom of the main peptide chain. The narrower lifetime distribution of the dansyl containing enkephalin analogues in more polar solvents (water and Me<sub>2</sub>SO) may be associated with the tendency of hydrophobic aromatic side chains (Dns and Trp) to sticking to the main peptide chain, which is less hydrophobic than the solvent.

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